

## Review for Ch. 8 Test

## 8.1 Describing Chemical Reactions

- know how to write word equations (i.e. you have be able to name compounds, including acids, ionic, and covalent compounds from Ch. 7)
- know how to write a formula equation
- What 2 things do coefficients represent in an equation? **Relative number of moles of a substance ; relative number of molecules needed for a reaction to occur**
- know common symbols used in a reaction
- know 4 evidences of a chemical reaction:
  1. **energy in the form of heat or light**
  2. **production of a gas**
  3. **change in color**
  4. **formation of a precipitate**
- know the Law of conservation of matter: **matter can neither be created nor destroyed in a chemical reaction (which means, mass of reactants must equal the mass of products)**
- know what a chemical equation represents: **relative amounts of reactants and products**
- know how to balance chemical equations (combustion reactions are particularly tricky, practice!)

1. Complete the equation and write a word equation for the following reaction:  $\text{ZnO} + \text{FeSO}_4 \rightleftharpoons$



**Zinc oxide reacts with iron (II) sulfate to yield zinc sulfate and iron (II) oxide**

2. Write a formula equation for the reaction of a solution of AgCl with potassium metal:



3. What elements are always diatomic gases at room temperature?

**All of the halogens (Group 17) ,  $\text{H}_2$ ,  $\text{N}_2$ ,  $\text{O}_2$**

4. Can you balance these reactions:



# Describing Chemical Reactions Section Review Answer Key

**Clifford Lane**



## **Describing Chemical Reactions Section Review Answer Key:**

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    - Chapter 2: Essential Elements of Describing Chemical Reactions Section Review Answer Key
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    - Chapter 4: Describing Chemical Reactions Section Review Answer Key in Specific Contexts
    - Chapter 5: Conclusion
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- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Describing Chemical Reactions Section Review Answer Key.

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