

## 11.1

## DESCRIBING CHEMICAL REACTIONS

## Section Review

## Objectives

- 1. Represent the processes of a reaction.
- 2. Explain how to write a balanced equation.
- 3. List the steps for writing a complete chemical equation.

## Vocabulary

- 1. chemical equation
- 2. chemical equation
- 3. reactant
- 4. coefficient
- 5. balanced equation

## Part A Completion

Write the coefficient in front of each reactant, product, and catalyst and balance the equation. Indicate the physical states of the substances using the symbols (s), (l), (g), and (aq).

1. A chemical equation is always represented by a balanced \_\_\_\_\_ 1 \_\_\_\_\_  
 The chemical formula for a chemical compound is \_\_\_\_\_ 2 \_\_\_\_\_ The mass  
 number for an element is represented by the \_\_\_\_\_ 3 \_\_\_\_\_ An element with the  
 mass number 40 is \_\_\_\_\_ 4 \_\_\_\_\_ A chemical equation is written with the  
 reactants on the left and the products on the right \_\_\_\_\_ 5 \_\_\_\_\_  
 coefficients are placed \_\_\_\_\_ 6 \_\_\_\_\_ in front of the chemical  
 formulas to show the ratio of the number of atoms of each \_\_\_\_\_ 7 \_\_\_\_\_ which is needed to  
 balance the equation and which is determined by the law of \_\_\_\_\_ 8 \_\_\_\_\_  
 the conservation of mass.

2. The chemical formula for water is \_\_\_\_\_ 9 \_\_\_\_\_  
 The chemical formula for hydrogen gas is \_\_\_\_\_ 10 \_\_\_\_\_  
 The chemical formula for oxygen gas is \_\_\_\_\_ 11 \_\_\_\_\_  
 The chemical formula for carbon dioxide is \_\_\_\_\_ 12 \_\_\_\_\_  
 The chemical formula for sodium chloride is \_\_\_\_\_ 13 \_\_\_\_\_  
 The chemical formula for calcium hydroxide is \_\_\_\_\_ 14 \_\_\_\_\_  
 The chemical formula for sulfuric acid is \_\_\_\_\_ 15 \_\_\_\_\_  
 The chemical formula for nitric acid is \_\_\_\_\_ 16 \_\_\_\_\_  
 The chemical formula for phosphoric acid is \_\_\_\_\_ 17 \_\_\_\_\_  
 The chemical formula for acetic acid is \_\_\_\_\_ 18 \_\_\_\_\_  
 The chemical formula for hydrochloric acid is \_\_\_\_\_ 19 \_\_\_\_\_  
 The chemical formula for hydrofluoric acid is \_\_\_\_\_ 20 \_\_\_\_\_  
 The chemical formula for boric acid is \_\_\_\_\_ 21 \_\_\_\_\_  
 The chemical formula for carbonic acid is \_\_\_\_\_ 22 \_\_\_\_\_  
 The chemical formula for formic acid is \_\_\_\_\_ 23 \_\_\_\_\_  
 The chemical formula for lactic acid is \_\_\_\_\_ 24 \_\_\_\_\_  
 The chemical formula for malic acid is \_\_\_\_\_ 25 \_\_\_\_\_  
 The chemical formula for tartaric acid is \_\_\_\_\_ 26 \_\_\_\_\_  
 The chemical formula for citric acid is \_\_\_\_\_ 27 \_\_\_\_\_  
 The chemical formula for ascorbic acid is \_\_\_\_\_ 28 \_\_\_\_\_  
 The chemical formula for folic acid is \_\_\_\_\_ 29 \_\_\_\_\_  
 The chemical formula for vitamin C is \_\_\_\_\_ 30 \_\_\_\_\_  
 The chemical formula for vitamin D is \_\_\_\_\_ 31 \_\_\_\_\_  
 The chemical formula for vitamin E is \_\_\_\_\_ 32 \_\_\_\_\_  
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 The chemical formula for vitamin E is \_\_\_\_\_ 50 \_\_\_\_\_  
 The chemical formula for vitamin K is \_\_\_\_\_ 51 \_\_\_\_\_  
 The chemical formula for vitamin A is \_\_\_\_\_ 52 \_\_\_\_\_  
 The chemical formula for vitamin B is \_\_\_\_\_ 53 \_\_\_\_\_  
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 The chemical formula for vitamin E is \_\_\_\_\_ 98 \_\_\_\_\_  
 The chemical formula for vitamin K is \_\_\_\_\_ 99 \_\_\_\_\_  
 The chemical formula for vitamin A is \_\_\_\_\_ 100 \_\_\_\_\_

1. balanced
2. chemical formula
3. mass number
4. calcium
5. products
6. right
7. coefficients
8. conservation of mass
9. H<sub>2</sub>O
10. H<sub>2</sub>
11. O<sub>2</sub>
12. CO<sub>2</sub>
13. NaCl
14. Ca(OH)<sub>2</sub>
15. H<sub>2</sub>SO<sub>4</sub>
16. HNO<sub>3</sub>
17. H<sub>3</sub>PO<sub>4</sub>
18. CH<sub>3</sub>COOH
19. HCl
20. HF
21. H<sub>3</sub>BO<sub>3</sub>
22. H<sub>2</sub>CO<sub>3</sub>
23. HCOOH
24. C<sub>3</sub>H<sub>7</sub>COOH
25. C<sub>4</sub>H<sub>6</sub>O<sub>6</sub>
26. C<sub>6</sub>H<sub>8</sub>O<sub>7</sub>
27. C<sub>10</sub>H<sub>16</sub>O<sub>6</sub>
28. C<sub>12</sub>H<sub>14</sub>O<sub>6</sub>
29. C<sub>15</sub>H<sub>22</sub>O<sub>6</sub>
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89. C<sub>315</sub>H<sub>726</sub>O<sub>6</sub>
90. C<sub>320</sub>H<sub>738</sub>O<sub>6</sub>
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# Describing Chemical Reactions 11 1 Section Review

**Lauren Gardner**



## Describing Chemical Reactions 11 1 Section Review:

**Principles Of Descriptive Inorganic Chemistry** Gary Wulfsberg, 1991-05-29 This is the only text currently available organized by class of compound and by property or reaction type not group by group or element by element which requires students to memorize isolated facts This is the only text currently available organized by class of compound and by property or reaction type not group by group or element by element which requires students to memorize isolated facts Translated into Italian [Class 8-12 Chemistry Questions and Answers PDF](#) Arshad Iqbal, The Class 8 12 Chemistry Quiz Questions and Answers PDF Grade 8 12 Chemistry Competitive Exam Questions Chapter 1 15 Practice Tests Chemistry Textbook Questions for Beginners includes Questions to solve problems with hundreds of class questions Class 8 12 Chemistry Questions and Answers PDF book covers basic concepts and analytical assessment tests Class 8 12 Chemistry Quiz PDF book helps to practice test questions from exam prep notes The Grade 8 12 Chemistry Quiz Questions and Answers PDF eBook includes Practice material with verbal quantitative and analytical past papers questions Class 8 12 Chemistry Questions and Answers PDF Free download chapter 1 a book to review textbook questions on chapters Molecular structure acids and bases atomic structure bonding chemical equations descriptive chemistry equilibrium systems gases laboratory liquids and solids mole concept oxidation reduction rates of reactions solutions thermochemistry Questions for high school and college revision questions Chemistry Interview Questions and Answers PDF Download free eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The Grade 8 12 Chemistry Interview Questions Chapter 1 15 PDF book includes high school workbook questions to practice Questions for exam Chemistry Practice Tests a textbook s revision guide with chapters Questions for NEET MCAT GRE GMAT SAT ACT competitive exam Grade 8 12 Chemistry Questions Bank Chapter 1 15 PDF book covers problem solving exam tests from chemistry practical and textbook s chapters as Chapter 1 Molecular Structure Questions Chapter 2 Acids and Bases Questions Chapter 3 Atomic Structure Questions Chapter 4 Bonding Questions Chapter 5 Chemical Equations Questions Chapter 6 Descriptive Chemistry Questions Chapter 7 Equilibrium Systems Questions Chapter 8 Gases Questions Chapter 9 Laboratory Questions Chapter 10 Liquids and Solids Questions Chapter 11 Mole Concept Questions Chapter 12 Oxidation Reduction Questions Chapter 13 Rates of Reactions Questions Chapter 14 Solutions Questions Chapter 15 Thermochemistry Questions The Molecular Structure Quiz Questions PDF e Book Chapter 1 interview questions and answers on polarity three dimensional molecular shapes The Acids and Bases Quiz Questions PDF e Book Chapter 2 interview questions and answers on Arrhenius concept Bronsted lowry concept indicators introduction Lewis concept pH strong and weak acids and bases The Atomic Structure Quiz Questions PDF e Book Chapter 3 interview questions and answers on electron configurations experimental evidence of atomic structure periodic trends quantum numbers and energy levels The Bonding Quiz Questions PDF e Book Chapter 4 interview questions and answers on ionic bond covalent bond dipole dipole forces hydrogen bonding intermolecular forces London dispersion forces

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Descriptive Inorganic Chemistry James E. House,Kathleen A. House,2015-09-10 House s Descriptive Inorganic Chemistry Third Edition provides thoroughly updated coverage of the synthesis reactions and properties of elements and inorganic compounds Ideal for the one semester ACS recommended sophomore or junior level course in descriptive inorganic chemistry this resource offers a readable and engaging survey of the broad spectrum of topics that deal with the preparation properties and use of inorganic materials Using rich graphics to enhance content and maximize learning the book covers the chemical behavior of the elements acid base chemistry coordination chemistry organometallic compounds and numerous other topics to provide a coherent treatment of the field The book pays special attention to key subjects such as chemical bonding and Buckminster Fullerenes and includes new and expanded coverage of active areas of research such as bioinorganic chemistry green chemistry redox chemistry nanostructures and more Highlights the Earth s crust as the source of most inorganic compounds and explains the transformations of those compounds into useful products Provides a coherent treatment of the field covering the chemical behavior of the elements acid base chemistry coordination chemistry and

organometallic compounds Connects key topics to real world industrial applications such as in the area of nanostructures Includes expanded coverage on bioinorganic chemistry green chemistry redox chemistry superacids catalysis and other areas of recent development      **Study Guide for Anatomy & Physiology - E-Book** Linda Swisher, Kevin T. Patton, 2014-12-02

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*Reviews* ,1974      Energy Research Abstracts ,1990      **Nuclear Science Abstracts** ,1975      **Scientific and Technical**

**Aerospace Reports** ,      Prentice Hall Chemistry ,2000      *Sif: Chemistry 5na Tb* J. G. R. Briggs, 2009      **Nano- and Biomaterials** Zhypargul Abdullaeva, 2017-06-21 A comprehensive introduction to nano and biomaterials shining light on the different research disciplines from various perspectives The straightforward and well structured concept is designed to cater for entrants as well as experienced researchers in the field of nanotechnology The initial chapters introduce nanomaterials their classification and synthesis techniques while subsequent chapters discuss the various characterization tools as well as mechanical properties and their applications in biotechnological and biomedical fields Further understanding of the topic is supported by case studies used for practical purposes The book concludes with a look at future technology advances With its explanation of a wide variety of materials this is an essential reference for chemists physicists materials scientists and biomedical engineers      **Chemical Reactivity in Confined Systems** Pratim Kumar Chattaraj, Debdu

Chakraborty, 2021-08-23 An insightful analysis of confined chemical systems for theoretical and experimental scientists Chemical Reactivity in Confined Systems Theory and Applications presents a theoretical basis for the molecular phenomena observed in confined spaces The book highlights state of the art theoretical and computational approaches with a focus on obtaining physically relevant clarification of the subject to enable the reader to build an appreciation of underlying chemical principles The book includes real world examples of confined systems that highlight how the reactivity of atoms and molecules change upon encapsulation Chapters include discussions on recent developments related to several host guest systems including cucurbit n uril ExBox 4 clathrate hydrates octa acid cavitand metal organic frameworks MOFs covalent

organic frameworks COFs zeolites fullerenes and carbon nanotubes Readers will learn how to carry out new calculations to understand the physicochemical behavior of confined quantum systems Topics covered include A thorough introduction to global reactivity descriptors including electronegativity hardness and electrophilicity An exploration of the Fukui function as well as dual descriptors higher order derivatives and reactivity through information theory A practical discussion of spin dependent reactivity and temperature dependent reactivity Concise treatments of population analysis reaction force electron localization functions and the solvent effect on reactivity Perfect for academic researchers and graduate students in theoretical and computational chemistry and confined chemical systems Chemical Reactivity in Confined Systems Theory and Applications will also earn a place in the libraries of professionals working in the areas of catalysis supramolecular chemistry and porous materials

Chemistry as a Game of Molecular Construction Sason Shaik, 2016-01-27 Chemistry as a Game of Molecular Construction The Bond Click Way utilizes an innovative and engaging approach to introduce students to the basic concepts and universal aspects of chemistry with an emphasis on molecules beauty and their importance in our lives Offers a unique approach that portrays chemistry as a window into mankind s material chemical essence Reveals the beauty of molecules through the click method a teaching methodology comprised of the process of constructing molecules from building blocks Styles molecular construction in a way that reveals the universal aspect of chemistry Allows students to construct molecules from the simple hydrogen molecule all the way to complex strands of DNA thereby showing the overarching unity of matter Provides problems sets and solutions for each chapter

**Inorganic Chemistry** Mark Weller, Mark T. Weller, Tina Overton, Jonathan Rourke, Fraser Armstrong, 2014 Leading the reader from the fundamental principles of inorganic chemistry right through to cutting edge research at the forefront of the subject Inorganic Chemistry Sixth Edition is the ideal course companion for the duration of a student s degree The authors have drawn upon their extensive teaching and research experience in updating this established text the sixth edition retains the much praised clarity of style and layout from previous editions while offering an enhanced Frontiers section Exciting new applications of inorganic chemistry have been added to this section in particular relating to materials chemistry and medicine This edition also sees a greater use of learning features to provide students with all the support they need for their studies Providing comprehensive coverage of inorganic chemistry while placing it in context this text will enable the reader to fully master this important subject Online Resource Centre For registered adopters of the text Figures marginal structures and tables of data ready to download Test bank For students Answers to self tests and exercises from the book Videos of chemical reactions Tables for group theory Web links Interactive structures and other resources on [www.chemtube3d.com](http://www.chemtube3d.com)

**The Saturday Review of Politics, Literature, Science and Art**, 1862 *Elements of the p-Block* C J Harding, D A Johnson, Rob Janes, 2007-10-31 Presenting a systematic approach to the chemistry of the p Block elements and hydrogen this book also introduces some basic topics concerning chemical bonding such as oxidation numbers bond strengths dipole moments and

intermolecular forces The chemistry is illustrated by coverage of the biological role of nitric oxide and of hydrogen bonding and the new chemistry of carbon nanotubes Applied aspects of the topic are developed in the two Case Studies which examine the causes and prevention of acid rain and the inorganic chemical industry The accompanying CD ROMs cover silicate mineral structures the inert pair effect and a database of chemical reactions of the p Block elements The Molecular World series provides an integrated introduction to all branches of chemistry for both students wishing to specialise and those wishing to gain a broad understanding of chemistry and its relevance to the everyday world and to other areas of science The books with their Case Studies and accompanying multi media interactive CD ROMs will also provide valuable resource material for teachers and lecturers The CD ROMs are designed for use on a PC running Windows 95 98 ME or 2000

**Shriver and Atkins' Inorganic Chemistry** Peter Atkins, 2010 Inorganic Chemistry fifth edition represents an integral part of a student's chemistry education Basic chemical principles are set out clearly in Foundations and are fully developed throughout the text culminating in the cutting edge research topics of the Frontiers which illustrate the dynamic nature of inorganic chemistry

**Sequence-Controlled Polymers** Jean-François Lutz, 2018-04-09 Edited by a leading authority in the field the first book on this important and emerging topic provides an overview of the latest trends in sequence controlled polymers Following a brief introduction the book goes on to discuss various synthetic approaches to sequence controlled polymers including template polymerization genetic engineering and solid phase chemistry Moreover monomer sequence regulation in classical polymerization techniques such as step growth polymerization living ionic polymerizations and controlled radical polymerizations are explained before concluding with a look at the future for sequence controlled polymers With its unique coverage of this interdisciplinary field the text will prove invaluable to polymer and environmental chemists as well as biochemists and bioengineers

**Medical Review of Reviews**, 1898

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