

- | Temp. | Experiment A | Experiment B |
|---|--|---|
| Independent Variable
<i>Volume</i> | # of particles | Temperature |
| Dependent Variable
<i>External & Internal Pressure</i> | Internal Pressure | Internal Pressure |
| Controlled Variable
<i># of particles</i> | External Pressure / Volume / Temperature | External Pressure / Volume / # of particles |

5. Of the variables that were controlled in both Experiment A and Experiment B in Model 1, one requires a nonflexible container. Name this variable, and explain why a nonflexible container is necessary. In your answer, consider the external and internal pressure data given in Model 1.

Experiment Requires a nonflexible container as the independent variable is # of particles which would normally expand a container. The second experiment does not require a nonflexible container as the # of particles does not change. \uparrow temp \uparrow Volume

Pressure is caused by molecules hitting the sides of a container or other objects. The pressure changes when the molecules change *how often* or *how hard* they hit. A nonflexible container is needed if the gas sample is going to have an internal pressure that is different from the external pressure. If a flexible container is used, the internal pressure and external pressure will always be the same because they are both



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