EXPERIMENT 5 ADSORPTION FROM SOLUTION

Introduction

The term adsorption is used to describe the fact that there is a greater concentration of the adsorbed molecules at the surface of the solid than in the bulk solution. In general, one uses solid adsorbents of small size and often with surface imperfections such as cracks and holes which serve to increase the surface area per unit mass greatly over the apparent geometrical area. Such small, porous particles may have specific areas in the range from 10 to 1000 m²g⁻¹. Some examples of adsorbents commonly used in experiments of this kind are charcoal, silica gel, alumina, zeolites, and molecular sieves. The adsorption from aqueous solutions of acetic cid on charcoal will be investigated in the present experiment.

The type of interaction between the adsorbed molecule and the solid surface varies over a wide range from weak nonpolar van der Waals' forces to strong chemical bonding. Examples of adsorption where ionic or covalent bonding occurs are the adsorption of chloride ions and silver chloride (ionic) or of oxygen gas on metals where oxygen-metal bonds are formed (covalent). In these cases, the process is called chemisorption, and it is generally characterized by high heats of adsorption (from 10 to 100 kcal mol⁻¹ of gas adsorbed). Chemisorption is highly specific in nature and depends on the chemical properties of both the surface molecules and the adsorbed molecules. Adsorption arising from the weaker van der Waals' and dipole forces is not so specific in character and can take place in any system at low or moderate temperatures. This type of adsorption is called physical adsorption and is usually associated with low heats of adsorption (less than about 10 keal mol'). Physical adsorption forces are similar to those which cause condensation of gases into liquid or solids. When an adsorbing molecule approaches the surface of the solid, there is an interaction between that molecule and the molecule in the surface which tends to concentrate the adsorbing molecules on the surface in much the same way that a gas molecule is condensed onto the surface of bulk liquid. Another respect in which physical adsorption is similar to liquid condensation is the fact that molar heats are of adsorption are of the same order of magnitude as molar heats of vaporization.

The amount adsorbed per gram of solid depends on the specific area of the solid, the equilibrium solute concentration in the solution (or pressure in the case of adsorption from the gas phase), the temperature, and the nature of the molecules involved. From measurements at constant temperature, one can obtain a plot of N, the number of moles adsorbed per gram of solid, versus c, the equilibrium solute concentration. This is called an adsorption isotherm.

Often it is possible to represent experimental results over a limited range by an empirical isotherm suggested by Freundlich:

$$N = Kc^n \tag{1}$$

Experiment 5 Adsorption From Solution

Zhe Xu

Experiment 5 Adsorption From Solution:

The Role of Iron Sulfides in Controlling Trace Heavy Metals in Anaerobic Sediments, 1977 The study of the kinetics of oxidation of the ferrous monosulfide mackinawite and the fate of the associated heavy metals are important to a better understanding of the processes and mechanisms controlling the release transport and retention of heavy metals in natural aquatic systems This project undertook a detailed study of several selected aspects of the aqueous chemistry of heavy metals and of the oxidative dissolution of FeS's and the fate of associated heavy metals both during and after the oxidation reaction The material is organized into sections dealing with the basic aqueous chemistry of iron and sulfur silver cadmium adsorption models experimental methodology results and discussion Research Reporting Series ,1975 Removal of Chromium from Plating Rinse Water Using Activated Carbon Richard B. Landrigan, J. B. Hallowell, 1975 Radiation Protection and Nuclear Safety: Experiment and Guidance Xiaogin Nie, Junling Wang, 2025-05-31 This book reviews the theory and practice on radiation protection and nuclear safety based on experimental case studies It concentrates on research advances in radiation detection radiation monitoring radioactive waste treatment and disposal It can both serve as a textbook for advanced undergraduates majoring in radiation protection and nuclear safety nuclear engineering and technology nuclear chemical engineering and nuclear fuel engineering and a reference for industry and institutes The basis of English translation of this book originally in Chinese was facilitated by artificial intelligence The content was later revised by the author for accuracy Proceedings of the First International Congress of Radiation Protection W. S. Snyder, 2013-09-12 Proceedings of the First International Congress of Radiation Protection Part 1 is a repository of research studies presented at the First International Congress of Radiation Protection sponsored by the International Radiation Protection Association IRPA held in Rome in September 1966 The compendium contains research papers from researchers all over the world covering every major subject category of radiation protection Subjects studied include development of health physics biological effects of radiation personal dosimetry methods health physics engineering and radiochemical analysis Discussions on environmental contamination fallout and food chains the medical aspects of radiation protection and decontamination of fission products on human skin and hair are presented as well The book offers insights and information that will be useful to scientists and workers in the field of atomic energy and nuclear research Organic Modification of Natural Clay Minerals and Its Adsorption on Anionic PPCPs Ken Sun, 2023-09-15 This book mainly investigates the adsorption of anionic diclofenac sodium on natural clay minerals the adsorption of anionic chloramphenicol on natural clay minerals and the behavior of surface and interface modification or adsorption on clay minerals It has significant guidance for learning the transfer and fate of pharmaceuticals and personal care products PPCPs in the soil It develops an efficient low cost adsorbent material and provides theoretical basis and valuable reference for the effective prevention of groundwater contamination of PPCPs in the field of environmental engineering Physical Chemistry: Experimental and Theoretical G. Van

Praagh,2015-11-19 Originally published in 1950 this textbook was intended for school students with the aim of providing an introductory understanding of chemistry The book introduces physical chemistry through multiple and diverse experiments each experiment designed to reinforce a new topic and reflect theorems approaches and historical development Notably the treatment throughout is from the point of view of the kinetic molecular theory rather than that of the laws of thermodynamics whilst emphasis is also placed upon physico chemical phenomena and their significance in various branches of science such as metallurgy chemical syntheses and mineralogy There are twelve chapters in total with chapter titles ranging from Atoms and molecules to Mass action and the ionic dissociation theory Various diagrams and plate sections are also included for reference This book will be of value to chemistry students and scholars as well as those interested in the history of education

Geological Survey Professional Paper ,1968 **Innovative Materials and Methods for Water Treatment** Marek Bryjak, Nalan Kabay, Bernabe L. Rivas, Jochen Bundschuh, 2016-02-17 Due to increasing demand for potable and irrigation water water suppliers have to use alternative resources They either have to regenerate wastewater or deal with contaminated surface water This book brings together the experiences of various experts in preparing of innovative materials that are selective for arsenic and chromium removal and in Security of Industrial Water Supply and Management Aysel T. Atimtay, Subhas K. Sikdar, 2011-08-24 Over time the increased use of fresh water for agriculture and industry together with contamination from discharges of pollutants mean that ever more areas of the planet are becoming water stressed Because of the competing needs of communities and industry for fresh water industry will be challenged to meet its growing demands for water which is essential for producing the goods and services that would boost human welfare Thus industry will need to learn how to cost effectively purify and recycle its wastewater for reuse ultimately approaching a net zero discharge condition The chapters in this book written by international experts treat the technical issues of such treatment and water management and also provide guidance on technologies either existing or in development that can potentially achieve the goal of recycle reuse The book will serve as a useful reference for academics government and industry Advances in Chemical Engineering III Lin Yu, Jian Wei Guo, Guo Bin Yi, Qian Yu, 2013-09-04 professionals alike Selected peer reviewed papers from the 3rd International Conference on Chemical Engineering and Advanced Materials CEAM 2013 July 6 7 2013 Guangzhou China Geological Survey Professional Paper Geological Survey (U.S.),1968

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