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Materiali per l'Ingegneria Civile ed Industriale

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apr 16 2017 mempunyai titik didih dan titik leleh yang sangat rendah jari jari atom gas mulia sangat kecil mempunyai elektron valensi 8 khusus untuk helium elektron valensinya 2 sudah stabil sehingga sukar membentuk senyawa dengan unsur lain dalam tidak pernah di temukan senyawa gas mulia gas mulia terdapat sebagai molekul monoatomik

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fgas mulia dan halogen sejarah gas mulia golongan gas mulia atau golongan viii a adalah unsure unsur yang memiliki delapan elektron valensi dengan konfigurasi elektronik terluar $ns^2 np^6$ unsur unsur tersebut adalah helium he neon

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oct 6 2023 gas mulia ditemukan di dekat golongan halogen dalam tabel periodik karena unsur gas mulia memiliki konfigurasi elektron yang penuh unsur unsur tersebut tidak reaktif dan senyawanya tidak dikenal akibatnya gas ini dikenal dengan gas inert

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may 18 2022 dikutip dari buku kimia unsur golongan gas mulia halogen alkali dan alkali tanah yang disusun oleh tantri ambarsari 2020 halogen adalah senyawa yang unsur unsurnya dapat bereaksi dengan logam untuk membentuk garam reaktivitas unsur halogen sangat besar sehingga unsur unsurnya sulit ditemukan di alam