Gas laws practice test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

		· ·			
	300	Pressure is the force per unit a. volume	2.0	545214257	
				length.	
		 b. surface area. 	d.	depth.	
-	2.	 Why does a can collapse when a vacuum pump removes air from the can? a. The inside and outside forces balance out and crush the can. 			
		 The unbalanced outside force from 	atmospheri	c pressure crushes the can.	
		c. The atmosphere exerts pressure on	the inside o	if the can and crushes it.	
		d. The vacuum pump creates a force t	The vacuum pump creates a force that crushes the can.		
	3.				
		 a. the atmospheric pressure is less than standard atmospheric pressure. 			
		b. the atmospheric pressure is greater	than standa	rd atmospheric pressure.	
		c. the atmospheric pressure is equal to	o standard a	tmospheric pressure.	
		d. the atmospheric pressure cannot be	determined	L in the second	
	4.	4. Convert the pressure 0.840 atm to mm Hg.			
		a. 365 mm Hg	0.7	638 mm Hg	
		b. 437 mm Hg	d.	780 mm Hg	
	5	Convert the pressure 1.30 atm to kPa.			
	550	a 2 kPa	C.	132 kPa	
		b. 115 kPa	d.	245 kPa	
	65	Standard temperature is exactly			
		a. 1000C.	6.0	00C	
		b. 2738C.	d.	0 K.	
	7.	Three samples of gas each exert 740, mm Hg in separate 2 L containers. What pressure do they exert if the are all placed in a single 2 L container?			
		a. 247 mm Hg	60	1.48 ×10 ³ mm Hg	
		b. 740 mm Hg	d.	2.22 ×10 ³ mm Hg	
	8	If the temperature remains constant, V	and P conce	sent the original volume and pressure, and P' and P'	
	300	If the temperature remains constant, V and P represent the original volume and pressure, and V and P represent the new volume and pressure, what is the mathematical expression for Boyle's law?			
		a PP = PP			
		b. VV' = PP'	March 1		
		v	700	$V = \frac{VP}{P}$	
	0	9. A sample of oxygen occupies 560, mL when the pressure is 800,00 mm Hg. At constant t			
	3630	volume does the gas occupy when the pressure decreases to 700.0 mm Hg?			
		a. 80.0 mL	pressure de		
		b 490 ml.	a	640 mL	
		THE STATE OF THE CONTRACT OF THE PARTY.			
_	1.0.				
		 a. its volume increases. 	C.		
		 b. its volume is unchanged. 	d.	its density decreases.	

Gas Laws Practice Test Multiple Choice

Kenneth Rideout, Jonathan Wolf

Gas Laws Practice Test Multiple Choice:

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