Question 2.19:

If one of the two electrons of a H_2 molecule is removed, we get a hydrogen molecular ion H_2^* . In the ground state of an H_2^* , the two protons are separated by roughly 1.5 Å, and the electron is roughly 1 Å from each proton. Determine the potential energy of the system. Specify your choice of the zero of potential energy. Answer 2.19:

Proton I

O Electron

The system of two protons and one electron is represented in the given figure.

Charge on proton 1, $q_1 = 1.6 \times 10^{-19} \, \text{C}$ Charge on proton 2, $q_2 = 1.6 \times 10^{-19} \, \text{C}$ Charge on electron, $q_3 = -1.6 \times 10^{-19} \, \text{C}$ Distance between protons 1 and 2, $d_1 = 1.5 \times 10^{-19} \, \text{m}$ Distance between proton 1 and electron, $d_2 = 1 \times 10^{-19} \, \text{m}$ Distance proton 2 and electron, $d_3 = 1 \times 10^{-19} \, \text{m}$ The potential energy at infinity is zero.

Potential energy of the system,

$$V = \frac{q_1q_1}{4\pi e_a d_1} + \frac{q_2q_1}{4\pi e_a d_1} + \frac{q_1q_1}{4\pi e_a d_1}$$

Substituting $\frac{1}{4\pi e_-} = 9 \times 10^+ \text{ N/m}^2 \text{ C}^{-2}$, we obtain

$$V = \frac{9 \times 10^{9} \times 10^{-19} \times 10^{-19}}{10^{-19}} \left[-(16)^{2} + \frac{(1.6)^{2}}{1.5} + -(1.6)^{2} \right]$$

$$= -30.7 \times 10^{-19} J$$

$$= -19.2 \text{ eV}$$

Therefore, the potential energy of the system is -19.2 eV.

Question 2.20:

Two charged conducting spheres of radii a and b are connected to each other by a wire. What is the ratio of electric fields at the surfaces of the two spheres? Use the result obtained to explain why charge density on the sharp and pointed ends of a conductor is higher than on its flatter portions.

Answer 2.20:

Let a be the radius of a sphere A, Q_A be the charge on the sphere, and C_A be the capacitance of the sphere. Let b be the radius of a sphere B, Q_B be the charge on the sphere, and C_B be the capacitance of the sphere. Since the two spheres are connected with a wire, their potential (V) will become equal.

Let E, be the electric field of sphere A and E, be the electric field of sphere B. Therefore, their ratio,

$$\frac{E_{z}}{E_{y}} = \frac{Q_{z}}{4\pi e_{u} \times a_{z}} \times \frac{b^{2} \times 4\pi e_{u}}{Q_{y}}$$

$$\frac{E_{z}}{E_{y}} = \frac{Q_{z}}{Q_{y}} \times \frac{b^{2}}{a^{2}} \qquad ... (1)$$
However, $\frac{Q_{z}}{Q_{y}} = \frac{C_{z}b^{2}}{C_{y}b^{2}}$
And, $\frac{C_{z}}{C_{y}} = \frac{a}{b}$

$$\therefore \frac{Q_{z}}{Q_{z}} = \frac{a}{b} \qquad ... (2)$$

Putting the value of (2) in (1), we obtain

$$\therefore \frac{E_A}{E_B} = \frac{\alpha}{b} \frac{b^2}{\alpha^2} = \frac{b}{\alpha}$$

Therefore, the ratio of electric fields at the surface is b/a.

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<u>Multidimensional NMR Methods for the Solution State</u> Gareth A. Morris, James W. Emsley, 2012-12-19 The content of this volume has been added to eMagRes formerly Encyclopedia of Magnetic Resonance the ultimate online resource for NMR and MRI The literature of multidimensional NMR began with the publication of three papers in 1975 then nine in 1976 and fifteen

in 1977 and now contains many tens of thousands of papers Any attempt to survey the field must therefore necessarily be very selective not to say partial In assembling this handbook the Editors have sought to provide both the new researcher and the established scientist with a solid foundation for the understanding of multidimensional NMR a representative if inevitably limited survey of its applications an authoritative account of classic techniques such as COSY NOESY and TOSCY and an account of the latest progress in the development of multidimensional techniques. This handbook is structured in four parts The first opens with an historical introduction to and a brief account of the practicalities and applications of multidimensional NMR methods followed by a definitive survey of their conceptual basis and a series of articles setting out the generic principles of methods for acquiring and processing multidimensional NMR data In the second part the main families of multidimensional techniques arranged in approximate order of increasing complexity are described in detail from simple I resolved spectroscopy through to the powerful heteronuclear 3D and 4D methods that now dominate the study of structural biology in solution The third part offers and illustrative selection from the very wide range of applications of multidimensional NMR methods including some of the most recent developments in protein NMR Finally the fourth part introduces the idea of multidimensional spectra containing non frequency dimensions in which properties such as diffusion and relaxation are correlated About EMR Handbooks eMagRes Handbooks The Encyclopedia of Magnetic Resonance up to 2012 and eMagRes from 2013 onward publish a wide range of online articles on all aspects of magnetic resonance in physics chemistry biology and medicine The existence of this large number of articles written by experts in various fields is enabling the publication of a series of EMR Handbooks eMagRes Handbooks on specific areas of NMR and MRI The chapters of each of these handbooks will comprise a carefully chosen selection of articles from eMagRes In consultation with the eMagRes Editorial Board the EMR Handbooks eMagRes Handbooks are coherently planned in advance by specially selected Editors and new articles are written together with updates of some already existing articles to give appropriate complete coverage The handbooks are intended to be of value and interest to research students postdoctoral fellows and other researchers learning about the scientific area in question and undertaking relevant experiments whether in academia or industry Have the content of this Handbook and the complete content of eMagRes at your fingertips Visit www wileyonlinelibrary com ref eMagRes View other eMagRes publications here **Equations of Membrane Biophysics N** Lakshminarayanaiah, 2013-09-03 Equations of Membrane Biophysics provides an introduction to the relevant principles of thermodynamics kinetics electricity surface chemistry electrochemistry and other mathematical theorems so that the quantitative aspects of membrane phenomena in model and biological systems could be described The book begins by introducing several phenomena that arise across membranes both artificial and biological when different driving forces act across them This is followed by separate chapters on thermodynamic principles related to properties of dilute aqueous electrolyte solutions along with a review of the principles of electrostatics electrochemical principles Fick s laws of diffusion

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