

Name: **Kyleen Mitchell**

Date: **11-5-20**

Student Exploration: Balancing Chemical Equations

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

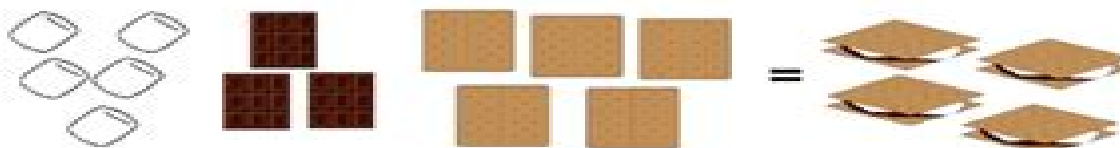
Vocabulary: coefficient, combustion, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript, synthesis

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

The scouts are making s'mores out of toasted marshmallows, chocolate, and graham crackers.

1. What is wrong with the image below?

What is wrong with the image below is that there are too many marshmallows and not enough chocolate and graham crackers.



2. Assuming a s'more requires two graham crackers, one marshmallow, and one piece of chocolate, how many s'mores could you make with the ingredients shown?

Assuming a s'more requires two graham crackers, one marshmallow, and one piece of chocolate, you could make 2 s'mores with the ingredients shown.

Gizmo Warm-up

In a chemical reaction, **reactants** interact to form **products**. This process is summarized by a chemical equation. In the *Balancing Chemical Equations* Gizmo, look at the floating molecules below the initial reaction: $\text{H}_2 + \text{O}_2 \rightarrow$ H_2O .

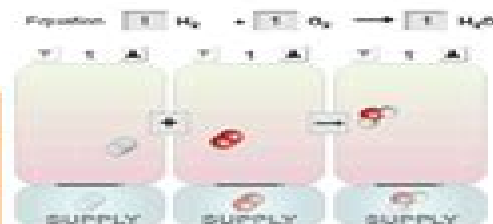
1. How many atoms are in a hydrogen **molecule** (H_2)?
2. How many atoms are in an oxygen molecule (O_2)?
3. How many hydrogen and oxygen atoms are in a water molecule (H_2O)?
4. In general, what does a **subscript** (such as the "2" in H_2) tell you about the molecule?

2

2

3

In general, a subscript like "2" in H_2 tells you that there are 2 H's in that molecule.



Balancing Chemical Equations Gizmo Worksheet Answers

N Colangelo



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Table of Contents Balancing Chemical Equations Gizmo Worksheet Answers

1. Understanding the eBook Balancing Chemical Equations Gizmo Worksheet Answers
 - The Rise of Digital Reading Balancing Chemical Equations Gizmo Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Balancing Chemical Equations Gizmo Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Balancing Chemical Equations Gizmo Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Balancing Chemical Equations Gizmo Worksheet Answers
 - Personalized Recommendations
 - Balancing Chemical Equations Gizmo Worksheet Answers User Reviews and Ratings

- Balancing Chemical Equations Gizmo Worksheet Answers and Bestseller Lists
- 5. Accessing Balancing Chemical Equations Gizmo Worksheet Answers Free and Paid eBooks
 - Balancing Chemical Equations Gizmo Worksheet Answers Public Domain eBooks
 - Balancing Chemical Equations Gizmo Worksheet Answers eBook Subscription Services
 - Balancing Chemical Equations Gizmo Worksheet Answers Budget-Friendly Options
- 6. Navigating Balancing Chemical Equations Gizmo Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Balancing Chemical Equations Gizmo Worksheet Answers Compatibility with Devices
 - Balancing Chemical Equations Gizmo Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Balancing Chemical Equations Gizmo Worksheet Answers
 - Highlighting and Note-Taking Balancing Chemical Equations Gizmo Worksheet Answers
 - Interactive Elements Balancing Chemical Equations Gizmo Worksheet Answers
- 8. Staying Engaged with Balancing Chemical Equations Gizmo Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Balancing Chemical Equations Gizmo Worksheet Answers
- 9. Balancing eBooks and Physical Books Balancing Chemical Equations Gizmo Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Balancing Chemical Equations Gizmo Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Balancing Chemical Equations Gizmo Worksheet Answers
 - Setting Reading Goals Balancing Chemical Equations Gizmo Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Balancing Chemical Equations Gizmo Worksheet Answers
 - Fact-Checking eBook Content of Balancing Chemical Equations Gizmo Worksheet Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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