

Experiment 1: Stereoisomerism: Preparation and Characterization

Subs. 234-100 (see Exp. 1, Sections 2.1-2.3, pp. 100-102)



The (experiment) equilibrium constant for the reaction above is:

$$K_c = \frac{\text{[2-butanone]} \times \text{[H}_2\text{O]}}{\text{[2-butanol]} \times \text{[PCC]}} = 1.0$$

Use the experimental procedure. The reaction is reversible. Use the following procedure to determine the concentration of each of the species in the equilibrium mixture (2-butanone is polarizable).

I. Exact Characterization (see Exp. 1, Sections 2.1-2.3, pp. 100-102)

Experimental composition:

	CH ₂ Cl ₂ (mL)	→	CH ₂ Cl ₂ (mL)	→	CH ₂ Cl ₂ (mL) + CH ₂ Cl ₂ (mL)	→	CH ₂ Cl ₂
2-butanol	1.0		1.0		2.0		2.0
2-butanone	0.0		0.0		0.0		0.0
2-butanol	1.0		0.0		1.0		1.0
2-butanone	0.0		0.0		0.0		0.0
2-butanol	1.0		0.0		1.0		1.0
2-butanone	0.0		0.0		0.0		0.0

Procedure: 2-butanol (see Exp. 1, Sections 2.1-2.3, pp. 100-102) and 2-butanone (see Exp. 1, Sections 2.1-2.3, pp. 100-102)

II. The 2-butanone for the Asymmetric 2-butanol (see Exp. 1, Sections 2.1-2.3, pp. 100-102)



Experiment 3 Ester Formation Preparation Of Benzocaine

Michal Rosen-Zvi



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