

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- _____ 1. Which statement is false?
- Ordinary chemical reactions do not involve changes in nuclei.
 - Atomic nuclei are very dense.
 - Nuclei are positively charged.
 - Electrons contribute only little to the mass of an atom.
 - The nucleus occupies nearly all the volume of an atom.
- _____ 2. The mass number of an atom is the number of _____ in the atom.
- protons
 - neutrons
 - protons plus the number of electrons
 - protons plus the number of neutrons
 - electrons plus the number of neutrons
- _____ 3. The atomic number of an atom is the number of _____ in the atom.
- protons plus the number of electrons
 - neutrons
 - protons
 - protons plus the number of neutrons
 - electrons plus the number of neutrons
- _____ 4. The atomic number of a certain element is 19, and its atomic weight is 39. An atom of the element contains _____ protons, _____ neutrons, and the chemical symbol for the element is _____.
- 19, 19, F
 - 19, 20, F
 - 19, 20, K
 - 20, 19, K
 - none of these
- _____ 5. An atom of the ^{23}Na isotope contains _____ electrons.
- 23
 - 12
 - 11
 - 10
 - 13
- _____ 6. A(*An*) _____ is a region of space in which there is a high probability of finding an electron in an atom.
- shell
 - atomic orbital
 - core
 - major energy level
 - nucleus
- _____ 7. What is the total number of orbitals in the fourth energy level ($n = 4$.)
- 4
 - 24
 - 16
 - 9
 - 18

General Chemistry Exam Questions And Answers

Alexander Smith



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