

① 7/12/18

This question paper contains 4 printed pages]

Roll No.

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S. No. of Question Paper : 39

Unique Paper Code : 32171101

Name of the Paper : Inorganic Chemistry—I

Name of the Course : B.Sc. (H) Chemistry

Semester : I

Duration : 3 Hours

Maximum Marks : 75



(Write your Roll No. on the top immediately on receipt of this question paper.)

Attempt six questions in all.

Question No. 1 is compulsory.

1. Explain any five of the following with suitable reason : 5×3

- (a) Which is more covalent : NaCl or NaI ?
- (b) Which has the greater bond dissociation energy :  $O_2$  or  $O_2^+$  ?
- (c) All the three N-O bonds in  $NO_3^-$  are equal.
- (d) Shape of  $dz^2$  orbital is different from other  $d$ -orbitals.
- (e)  $BeCl_2$  has zero dipole moment while  $H_2S$  has some value.
- (f) Which has greater melting point : *o*-nitrophenol or *p*-nitrophenol ?

P.T.O.

# Diploma First Semester Chemistry Questions Paper

**Y Pai**



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