



Determination Of % Available Chlorine In Bleaching Powder

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Aim

To determine the available chlorine in the given sample of bleaching powder by the iodometric method.



Introduction

- ❖ Bleaching powder, also known as chlorinated lime
- ❖ yellowish-white powder having a smell of chlorine and is readily soluble in water
- ❖ It is prepared by passing chlorine gas over slaked lime at a temperature of 35-45° and consists of a mixture of calcium hypochlorite $\text{Ca}(\text{OCl}_2)$ and calcium chloride CaCl_2 , in addition some amount of free slaked lime i.e. $\text{Ca}(\text{OH})_2 \cdot \text{H}_2\text{O}$ is generally present
- ❖ $\text{Ca}(\text{OCl}_2)$ is responsible for the bleaching action of bleaching powder. On treatment with glacial acetic acid, it liberates chlorine gas (Cl_2) as per the following reaction.



- ❖ The amount of chlorine liberated by the action of an acid on bleaching powder (CaOCl_2) is termed as available chlorine. The chlorine content of bleaching powder varies from 35 – 40%.



- ❖ bleaching powder also have strong germicidal and disinfectant properties.
- ❖ It finds application as a disinfectant for drinking water or swimming pool water.

Determination Of Available Chlorine In Bleaching Solution

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Gesellschaft in Zürich, 1861 **Laboratory Methods of the United States Army** United States. Surgeon-General's Office, 1919 *Chemistry: for B.Sc. Students Semester II (NEP-2020 KSHEC)* R L Madan, This textbook has been conceptualised for universities of Karnataka as per the recommended National Education Policy NEP 2020 to meet the needs of B Sc students of Chemistry Maintaining the traditional approach to the subject this textbook lucidly explains the basics of Inorganic and Physical Chemistry Important topics such as atomic structure periodicity of elements kinetic theory of gases liquid state liquid crystals and types of solids are aptly discussed to give an overview of inorganic and physical chemistry Laboratory work 33 experiments has also been included to help students achieve sound conceptual understanding and learn experimental procedures

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