



Lab Data



*Verify your calculation. In your set-up, do your units cancel?
Did you report your data to the correct number of significant figures?*

Moles of magnesium (mol) 0.00062

Temperature of water (°C) 22.0

Temperature of water (K) 295.15

Vapor pressure of water (mmHg) 19.8

Barometric pressure (mmHg) 770.0

Observations

There is bubble formation

Volume of hydrogen gas collected (mL) 15.00

Volume of hydrogen gas collected (L)

Pressure of hydrogen gas (mmHg)

Pressure of hydrogen gas (atm)

Moles of hydrogen gas produced (mol) 0.00062

Ideal gas constant ($\text{L}\cdot\text{atm}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$)



[How to calculate ideal gas constant](#)



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Ideal Gas Constant Lab 38 Answers

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Metals Abstracts, 1989 Environmental Bioenergetics Anke Marianne Herrmann, Doug LaRowe, Alain F. Plante, 2020-01-10 Gmelin Handbuch der anorganischen Chemie, 1982 *Excerpta Medica*, 1984 **Japanese Technical Abstracts**, 1987 **Physics Briefs**, 1991 **Research and Development Abstracts of the USAEC** U.S. Atomic Energy Commission. Division of Technical Information, 1962 **Proceedings of ICE-SEAM 2021: Special Edition** Mohd Fadzli Bin Abdollah, 2022-04-01 This e book is a compilation of papers presented at the 7th International Conference and Exhibition on Sustainable Energy and Advanced Materials ICE SEAM 2021 Virtual Platform Malaysia on 23 November 2021 This special edition of proceedings has 17 selected papers that focus on IR4.0 including 3D printing and advanced materials and how it might impact energy systems in numerous ways for sustainable development especially during the pandemic COVID19 **Research and Development Abstracts of the USAEC** U.S. Atomic Energy Commission, 1962-10
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