

Acids and Bases Worksheet

 Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases	
Conductivity in Aqueous Solution	very	very	
Taste	Sove	Bitter	
Feel		Slippery	
Reaction with Metal	produces H2 gas	-	
Reaction with Carbonate	produces Cb2 gas		
Colour on Litmus Paper	Red	Blue.	
Colour with Universal Indicator	Red	Purple.	
Colour with phenolphthalein	Colourless	Dink.	

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO ₃	A	nitric acid
Mg(OH)₂	_B_	magnesium hydroxide
H ₃ PO ₄	A	Phosphoric acid
H ₂ CO ₃	A	ourbonic acid
HF	A	hydrofluoric acid
Fe(OH)₃	B	iron (111) hydroxide.

3. Write the formula for each of the following acids/bases:

hydrochloric acid	HCI_	hydrochloric acid	HCI_
aluminum hydroxide	AI(OH)3	sodium hydroxide	NaOH
lithium hydroxide	LIOH	calcium hydroxide	Ca(04)2
sulfuric acid	H2504	sulfurous acid	H2503

Acids Bases Section 1 Answers

RS Peters



Decoding Acids Bases Section 1 Answers: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Acids Bases Section 1 Answers**," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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