

## ACIDS, BASES, & THE pH SCALE

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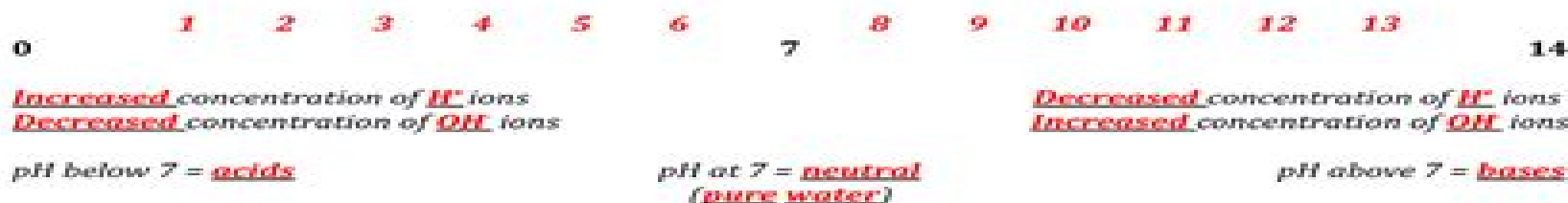
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### PART 1: ACIDS AND BASES

Comparing Acids and Bases	
Acids	Bases (Alkalines)
<ul style="list-style-type: none"><li>• <b>Sour</b> taste, leaves <b>burning</b> sensation</li><li>• <b>Increase</b> hydrogen ions (<math>H^+</math>) in solution</li><li>• Example: <b>HCl</b>     <b><math>H^+ + Cl^-</math></b></li><li>• Turns <b>litmus</b> (pH) paper to <b>red</b></li></ul>	<ul style="list-style-type: none"><li>• Sharp <b>bitter</b> taste, feels <b>slippery</b> to touch</li><li>• <b>Decrease</b> hydrogen ions (<math>H^+</math>) in solution (<b>Increase</b> hydroxide ions (<math>OH^-</math>) in solution)</li><li>• Example: <b>NaOH</b>     <b><math>Na^+ + OH^-</math></b></li><li>• Turns litmus (pH) paper to <b>blue</b></li></ul>

### PART 2: THE pH SCALE

- The **pH Scale** is a **measurement** of how acidic or how basic a solution is.
- The scale goes from **0** to **14**.
- **Compounds** are **acidic** if they have a pH **lower** than **7**.
- Compounds with a pH **higher** than **7** are **basic** or **alkaline**.
- Compounds with a pH **of 7**, are **neutral**.



### PART 3: BUFFERS

- A **buffer** is a **weak acid** or **base** that **prevents** sudden **changes** in pH.
  - Maintains **homeostasis**
- A buffer works by **binding** to  $H^+$  ions **releasing**  $H^+$  ions when the solution pH changes.
- Buffers **bind** to  $H^+$  ions when the **concentration** of  $H^+$  ions in solution **increases**.
- **NaOH**      **$Na^+ + OH^-$**  (The  $OH^-$  will bind to  $H^+$  ions to form  $H_2O$  – **neutralization**)
- Buffers **release**  $H^+$  ions when the **concentration** of  $H^+$  ions in solution **decreases**.  
**HCl**      **$H^+ + Cl^-$**

# Acids And Bases Study Answer Key

**TD Snyder**



## **Acids And Bases Study Answer Key:**

## **Acids And Bases Study Answer Key** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Acids And Bases Study Answer Key**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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