

## Solutions For Homework #1

### Problem 1:

$c$  is the speed of light,  $\lambda$  is the wavelength and  $f$  is the frequency. The equation  $c = \lambda \cdot f$  can be used to calculate one of the other values, if the other values are given. The speed of light is defined to be exactly 299792458 m/s or  $3 \cdot 10^8$  m/s. Thus, for a frequency of  $f = 68.245 \cdot 10^6 \text{ s}^{-1} = 68.245 \text{ MHz}$ ,

$$\lambda = \frac{c}{f} = \frac{299792458 \text{ m/s}}{68.245 \cdot 10^6 \text{ s}^{-1}} = \frac{3 \cdot 10^8 \text{ m/s}}{68.245 \cdot 10^6 \text{ s}^{-1}} = 4.404 \text{ m} \approx 4.4 \text{ m}$$

and for a wavelength of  $\lambda = 22.5 \text{ cm} = 0.225 \text{ m}$

$$f = \frac{c}{\lambda} = \frac{299792458 \text{ m/s}}{0.225 \text{ m}} = 1.3 \cdot 10^9 \text{ Hz} = 1.3 \text{ GHz}$$

### Problem 2:

The boiling temperature of water at sea level is 100°C. To convert this to degrees Kelvin, just add 273.15.

$$100^\circ\text{C} = 273.15 + 100^\circ\text{C} = 373.15\text{K}$$

### Problem 3:

The formula for the wavelength of the energy collision peak is

$$\lambda_{\text{max}} = \frac{2.48 \cdot 10^{-6} \text{ m}}{T}$$

where  $T$  is the black-body temperature. Using the temperatures for the surface of the sun ( $T = 5778\text{K}$ ) and an effective value for the earth ( $T = 255\text{K}$ ), we get

sun:	41.44 $\mu\text{m}$
earth:	1.0 $\mu\text{m}$

# Homework 1 Solutions Stanford Department

**David Baud**



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