

TWELFTH
EDITION

BECOMING *Aware*

A TEXT/WORKBOOK FOR
HUMAN RELATIONS &
PERSONAL ADJUSTMENT

■ ■ ■ Velma Walker ■ ■ ■



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G Psacharopoulos



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Getting Culture Regan A. R. Gurung, Loreto R. Prieto, 2023-07-03 How do we educate our students about cultural diversity and cultural differences and eliminate cultural ignorance stereotyping and prejudice What are the conceptual issues involved in reaching this goal How can we integrate these perspectives in disciplinary and diversity courses and the curriculum This book is a resource for answering these questions Within the framework of current scholarship and discussion of essential concepts it offers practical techniques and empirically proven best practices for teaching about diversity The book opens with a conceptual framework covering such issues as distinguishing teaching to a diverse audience from teaching about diversity and contrasting the incorporation of culture across the curriculum with tokenistic approaches Subsequent chapters identify classroom practices that can optimize students learning especially those from culturally diverse backgrounds describe feminist principles of education that promote learning for all students and address principles of effective on line instruction for diverse populations The book is intended for faculty integrating diversity into existing courses and for anyone creating courses on diversity The ideas and suggestions in the text can be incorporated into any class that includes a discussion of diversity issues or has a diverse student enrollment The contributors offer pragmatic and tested ways of overcoming student misconceptions and resistance and for managing emotional responses that can be aroused by the discussion of diversity The editors aim to stimulate readers thinking and inspire fresh ideas The book further provides teachers of diversity with a range of effective exercises and attends to such issues as teacher stress and burnout This book can also serve to inform and guide department chairs and other administrators in the design and implementation of diversity initiatives

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction especially nitration of methyl benzoate to give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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