Name	Date	Period
:- 1		

Double-Replacement Precipitation Lab

<u>Introduction</u>-Double-Replacement reactions occur when two ionic solutions mix to form an insoluble solid. If all cations and anions are soluble then no reaction will occur. During the lab procedure, several ionic solutions will be mixed to determine if a precipitate reaction will occur. Those reaction that do not form a precipitate will be marked for No Reaction (NR). Students will practice writing and balancing double-replacement reactions.

Pre-Lab questions:

- How would you describe a double replacement reaction to a friend who is struggling with reaction types?
- 2. What is a precipitate?

Minterials:

- · 0.2M sodium beomide, NaBr
- 0.2M sodium carbonate, Na₂CO₃
- 0.2M sodium phosphate, Na₃PO₄
- 0.2M sodium sulfate, Na₂SO₄
- · 0.2M sodium hydroxide, NaOH
- Transparency paper

- · 0.2M lithium chloride, LiCl.
- 0.2M copper(II) chloride, CuCl₂
- 0.2M zine chloride, ZnCl₂
- . 0.2M strontium chloride, SrCl₂
- 0.2M barium chloride BaCl₂
- · Background paper Template

Safety: You MUST wear goggles during this lab activity

Procedure:

- Place the overhead sheet over the background paper.
- Using the Data Table as a guide, mix every possible pair of solutions in a separate square of the background paper. <u>Use I or 2 drops of each solution</u>. Do NOT contaminate the individual pipets with different solutions.
- 3. Mix the solutions with a rinsed toothpick if you are uncertain if a precipitate formed. Rinse the toothpick after EACH mixing be careful of which end of the rod you use so that you do not touch the chemicals or contaminate your samples!
- 4. Observe each mixture carefully for signs of a precipitate. (Some precipitates are light in color and hard to see so you may want to slide the transparency over the black lab top on occasion.) Record your observations (presence or absence of a precipitate, and other characteristics such as color, texture, etc.) in the Data Table. If nothing happens, write NR for no reaction.
- Rinse and wige the transparency sheet into the sink. Dry the sheet.

Observations: Write the observations you made from combining these materials in the spaces in the table below. Note any color changes or precipitate (cloudiness/solid). If no change is observed write NR for no reaction.

	NaBr	Na ₂ CO ₃	NaoPO ₄	NagSO ₄	NaOH
LiCI					
CuClz					
ZnCl ₂					
SrCl ₂					
BaCl ₂					

Double Replacement Reaction Lab Answers

Leo J. Malone, Theodore Dolter

Double Replacement Reaction Lab Answers:

CHEMISTRY EXPERIMENTS James Signorelli, 2014-09-19 Gifted and talented students and any student interested in pursuing a science major in college needs a rigorous program to prepare them while they are still in high school This book utilizes a format where the application of several disciplines science math and language arts principles are mandated Each lab concludes with either an essay or a detailed analysis of what happened and why it happened This format is based on the expectations of joining a university program or becoming an industrial science professional The ideal student lab report would be written in a lab research notebook and then the essay or final analysis is done on a word processor to allow for repeat editing and corrections. The research notebook has all graph pages a title section and a place for the students and their assistants to sign and witness that exercise The basic mechanics of the lab report title purpose procedure diagrams data table math and calculations observations and graphs are handwritten into the book The conclusion is done on a word processor MS Word which allows the instructor to guide the student in writing and editing a complete essay using the MLA format When the final copy is completed the essay is printed and inserted into the lab notebook for grading At the end of the term the student has all their labs in one place for future reference These lab notebooks can be obtained for as little as 3 00 per book This is money well spent In our district the Board of Education buys the books for each student The BOE sees these books as expendable but necessary materials for all science and engineering instruction 40 Low-Waste, Low-Risk Chemistry Labs David Dougan, 1997 Builds essential process and thinking skills Investigates central chemistry concepts Features procedures for purchase storage use and disposal of chemicals Basic Concepts of Chemistry Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron s Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major

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