

CHEM 141L GENERAL CHEMISTRY-I**REVIEW EXAM #2**

(CH, 4-5-6; Masterton, Hurley, and North, 7th Ed.)
ECC

1. If aqueous solutions of nickel(II) chloride and potassium phosphate are mixed, which insoluble compound will form?
- $\text{Ni}_3(\text{PO}_4)_2$
 - NiCl_2
 - KCl
 - K_2Ni
 - K_3PO_4

ANS: A

2. Which of the following ionic compounds is not soluble in water?
- K_2CO_3
 - AgNO_3
 - CuBr_2
 - CaI_2
 - PbI_2

ANS: E

3. What is the net ionic equation for the reaction of aqueous sodium hydroxide and aqueous iron(II) chloride?
- $\text{Na}^+(\text{aq}) + \text{OH}^-(\text{aq}) \rightarrow \text{NaOH}(\text{s})$
 - $\text{Na}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{NaCl}(\text{s})$
 - $\text{Fe}^{2+}(\text{aq}) + 2\text{OH}^-(\text{aq}) \rightarrow \text{Fe}(\text{OH})_2(\text{s})$
 - $\text{Fe}^{2+}(\text{aq}) + \text{OH}^-(\text{aq}) \rightarrow \text{FeOH}(\text{s})$
 - $\text{Fe}^{2+}(\text{aq}) + 2\text{Cl}^-(\text{aq}) \rightarrow \text{FeCl}_2(\text{s})$

ANS: C

4. Write a balanced net ionic equation for the reaction of aqueous solutions of lead(II) nitrate and potassium chloride.
- $\text{Pb}(\text{NO}_3)_2(\text{aq}) + 2\text{KCl}(\text{aq}) \rightarrow \text{PbCl}_2(\text{s}) + 2\text{KNO}_3(\text{aq})$
 - $\text{Pb}^{2+}(\text{aq}) + 2\text{K}^+(\text{aq}) \rightarrow \text{PbK}_2(\text{s})$
 - $\text{Pb}^{2+}(\text{aq}) + 2\text{Cl}^-(\text{aq}) \rightarrow \text{PbCl}_2(\text{s})$
 - $\text{NO}_3^-(\text{aq}) + \text{K}^+(\text{aq}) \rightarrow \text{KNO}_3(\text{s})$
 - no precipitation occurs.

ANS: C

5. Which list contains only strong acids?
- HCl , HNO_3 , HF , HClO_4
 - H_2SO_4 , H_3PO_4 , HClO_4 , NH_3
 - HCl , HNO_3 , H_2PO_4 , HClO_3
 - HCl , H_2SO_4 , HClO_4 , HI
 - HNO_3 , H_2SO_4 , NaOH , H_2PO_4

Sample Exam 1 Name Chem 21

Marcel A. Müller



Sample Exam 1 Name Chem 21:

Chemical Analyses of Water Samples and Geophysical Logs from Cored Test Holes Drilled in the Central Oklahoma Aquifer, Oklahoma Jamie L. Schlottmann, Ron A. Funkhouser, 1991 *Exam Scorer Science (Chemistry) - Class XI (Chapterwise MCQs with 5 solved Model Papers for 2020 EXAM)* SBPD Editorial Board, 2019-10-03 An excellent book for Science students appearing in competitive professional and other examinations 1 Chemistry 2 5 Model Papers with OMR Sheet 3 Examination Paper JEE Advanced Chemistry - Unit wise Practice Test Papers Career Point Kota, 2020-07-20

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Table of Contents Sample Exam 1 Name Chem 21

1. Understanding the eBook Sample Exam 1 Name Chem 21
 - The Rise of Digital Reading Sample Exam 1 Name Chem 21
 - Advantages of eBooks Over Traditional Books
2. Identifying Sample Exam 1 Name Chem 21
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sample Exam 1 Name Chem 21
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Exam 1 Name Chem 21
 - Personalized Recommendations
 - Sample Exam 1 Name Chem 21 User Reviews and Ratings
 - Sample Exam 1 Name Chem 21 and Bestseller Lists
5. Accessing Sample Exam 1 Name Chem 21 Free and Paid eBooks
 - Sample Exam 1 Name Chem 21 Public Domain eBooks
 - Sample Exam 1 Name Chem 21 eBook Subscription Services
 - Sample Exam 1 Name Chem 21 Budget-Friendly Options
6. Navigating Sample Exam 1 Name Chem 21 eBook Formats
 - ePub, PDF, MOBI, and More
 - Sample Exam 1 Name Chem 21 Compatibility with Devices
 - Sample Exam 1 Name Chem 21 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sample Exam 1 Name Chem 21
 - Highlighting and Note-Taking Sample Exam 1 Name Chem 21
 - Interactive Elements Sample Exam 1 Name Chem 21
8. Staying Engaged with Sample Exam 1 Name Chem 21

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Sample Exam 1 Name Chem 21
- 9. Balancing eBooks and Physical Books Sample Exam 1 Name Chem 21
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sample Exam 1 Name Chem 21
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Sample Exam 1 Name Chem 21
 - Setting Reading Goals Sample Exam 1 Name Chem 21
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Sample Exam 1 Name Chem 21
 - Fact-Checking eBook Content of Sample Exam 1 Name Chem 21
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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