

**Multiple Choice**

Identify the letter of the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 1. Which statement is false?
- Ordinary chemical reactions do not involve changes in nuclei.
  - Atomic nuclei are very dense.
  - Nuclei are positively charged.
  - Electrons contribute only little to the mass of an atom.
  - The nucleus occupies nearly all the volume of an atom.
- \_\_\_\_\_ 2. The mass number of an atom is the number of \_\_\_\_\_ in the atom.
- protons
  - neutrons
  - protons plus the number of electrons
  - protons plus the number of neutrons
  - electrons plus the number of neutrons
- \_\_\_\_\_ 3. The atomic number of an atom is the number of \_\_\_\_\_ in the atom.
- protons plus the number of electrons
  - neutrons
  - protons
  - protons plus the number of neutrons
  - electrons plus the number of neutrons
- \_\_\_\_\_ 4. The atomic number of a certain element is 19, and its atomic weight is 39. An atom of the element contains \_\_\_\_\_ protons, \_\_\_\_\_ neutrons, and the chemical symbol for the element is \_\_\_\_\_.
- 19, 19, F
  - 19, 20, F
  - 19, 20, K
  - 20, 19, K
  - none of these
- \_\_\_\_\_ 5. An atom of the  $^{23}\text{Na}$  isotope contains \_\_\_\_\_ electrons.
- 23
  - 12
  - 11
  - 10
  - 13
- \_\_\_\_\_ 6. A(*An*) \_\_\_\_\_ is a region of space in which there is a high probability of finding an electron in an atom.
- shell
  - atomic orbital
  - core
  - major energy level
  - nucleus
- \_\_\_\_\_ 7. What is the total number of orbitals in the fourth energy level ( $n = 4$ .)
- 4
  - 24
  - 16
  - 9
  - 18

# General Chemistry Exam Questions And Answers

**Peter J. Mikulecky, Michelle Rose  
Gilman, Kate Brutlag**



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## **Table of Contents General Chemistry Exam Questions And Answers**

1. Understanding the eBook General Chemistry Exam Questions And Answers
  - The Rise of Digital Reading General Chemistry Exam Questions And Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying General Chemistry Exam Questions And Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an General Chemistry Exam Questions And Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from General Chemistry Exam Questions And Answers
  - Personalized Recommendations
  - General Chemistry Exam Questions And Answers User Reviews and Ratings

- General Chemistry Exam Questions And Answers and Bestseller Lists
- 5. Accessing General Chemistry Exam Questions And Answers Free and Paid eBooks
  - General Chemistry Exam Questions And Answers Public Domain eBooks
  - General Chemistry Exam Questions And Answers eBook Subscription Services
  - General Chemistry Exam Questions And Answers Budget-Friendly Options
- 6. Navigating General Chemistry Exam Questions And Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - General Chemistry Exam Questions And Answers Compatibility with Devices
  - General Chemistry Exam Questions And Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of General Chemistry Exam Questions And Answers
  - Highlighting and Note-Taking General Chemistry Exam Questions And Answers
  - Interactive Elements General Chemistry Exam Questions And Answers
- 8. Staying Engaged with General Chemistry Exam Questions And Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers General Chemistry Exam Questions And Answers
- 9. Balancing eBooks and Physical Books General Chemistry Exam Questions And Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection General Chemistry Exam Questions And Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine General Chemistry Exam Questions And Answers
  - Setting Reading Goals General Chemistry Exam Questions And Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of General Chemistry Exam Questions And Answers
  - Fact-Checking eBook Content of General Chemistry Exam Questions And Answers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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