

**Multiple Choice**

Identify the letter of the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 1. Which statement is false?
- Ordinary chemical reactions do not involve changes in nuclei.
  - Atomic nuclei are very dense.
  - Nuclei are positively charged.
  - Electrons contribute only little to the mass of an atom.
  - The nucleus occupies nearly all the volume of an atom.
- \_\_\_\_\_ 2. The mass number of an atom is the number of \_\_\_\_\_ in the atom.
- protons
  - neutrons
  - protons plus the number of electrons
  - protons plus the number of neutrons
  - electrons plus the number of neutrons
- \_\_\_\_\_ 3. The atomic number of an atom is the number of \_\_\_\_\_ in the atom.
- protons plus the number of electrons
  - neutrons
  - protons
  - protons plus the number of neutrons
  - electrons plus the number of neutrons
- \_\_\_\_\_ 4. The atomic number of a certain element is 19, and its atomic weight is 39. An atom of the element contains \_\_\_\_\_ protons, \_\_\_\_\_ neutrons, and the chemical symbol for the element is \_\_\_\_\_.
- 19, 19, F
  - 19, 20, F
  - 19, 20, K
  - 20, 19, K
  - none of these
- \_\_\_\_\_ 5. An atom of the  $^{23}\text{Na}$  isotope contains \_\_\_\_\_ electrons.
- 23
  - 12
  - 11
  - 10
  - 13
- \_\_\_\_\_ 6. A(*An*) \_\_\_\_\_ is a region of space in which there is a high probability of finding an electron in an atom.
- shell
  - atomic orbital
  - core
  - major energy level
  - nucleus
- \_\_\_\_\_ 7. What is the total number of orbitals in the fourth energy level ( $n = 4$ .)
- 4
  - 24
  - 16
  - 9
  - 18

# General Chemistry Exam Questions And Answers

**Donovan Young**



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