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Assignment #1 – Compound Names and Formulas Single-valent ions only

A. Name these Compounds 1. Li₂S <u>LHNiv/m Sulphide</u>	10. GeF4 Greymonium Fluoride
2. Cao Calcium Oxide	
3. NaF Sodium Fluoride	
4. CaBra Calcium Bromide	
5. MgCl2 Magnesium Chloride	
6. BBr. Boran Bramide	
7. Cs20 Ctsium Oxide	
8. FIBIT Francium Bramide	
9. Agos Silver Sulphyle	The state of the s
1. sodium chloride NaCl	
	7 - 14 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -
2. magnesium fluoride MgF ₂	
3. silver oxide <u>VAg_O</u>	
4. indium bromide INBYs	14. potassium oxide
5. zinc bromide ZNBY2	15. beryllium sulphide BCS
6. neodymium oxide <u>NdzO3</u>	16. lithium sulphide
7. thorium sulphide TNS2	17. hydrogen bromide HBY
8. actinium oxide <u>AC203</u>	18. strontium nitride SY3N2
Oak-	
9. radium bromide ROBY2	19. calcium oxideCO\O

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